



**RX-003-001646**

Seat No. \_\_\_\_\_

**B. Sc. (Sem. VI) (CBCS) Examination**

**March - 2019**

**IC - 601 : Dyes-2 & Polymer Technology**

**Faculty Code : 003**

**Subject Code : 001646**

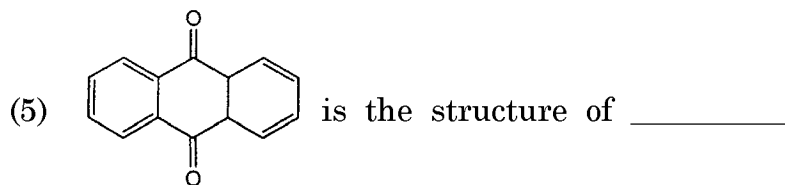
Time :  $2\frac{1}{2}$  Hours]

[Total Marks : 70

- Instructions :**
- (1) All the questions are compulsory
  - (2) Figures to the right indicate maximum marks.
  - (3) Draw labeled diagram wherever necessary.
  - (4) Assume suitable data.
  - (5) Question-1 carries 20 marks.
  - (6) Question-2 & 3 carry 25 marks each.

**1 Answer the following questions : 20**

- (1) In azo dye, D stands for what?
- (2)  $A \rightarrow Z \leftarrow A'$  is a type of bis azo dye. True/False?
- (3) Give one example of mono azo dye.
- (4) Give the structure of  $\beta$ -naphthol.



- (6) Sulfonation of naphthalene at low temperature gives \_\_\_\_\_.
- (7) Give full form of HPLC.
- (8) The ratio of distance travelled by solute to solvent is known as?
- (9) Give one example of stationary phase used in TLC.
- (10) Give full form of EBT.

- (11) \_\_\_\_\_ can be used as monomer for addition polymerization.
- (12) Monomer having \_\_\_\_\_ functional groups can create branched polymer.
- (13) Give structure of ethane 1,2-diol.
- (14) \_\_\_\_\_ is catalyst for manufacturing of stereo regular polymer.
- (15) Give raw materials for manufacturing of SBR polymer.
- (16) \_\_\_\_\_ can be used as initiator for free radical polymerization.
- (17) \_\_\_\_\_ is commonly called Buna-S.
- (18) "Polythene is thermosetting polymer" is this statement true or false?
- (19) \_\_\_\_\_ is full form of LDPE.
- (20) Give formula of phosgene.

2 (A) Answer any **Three** : **6**

- (1) What is weight average molecular weight?
- (2) Define : (a) Degree of polymerization, (b) Monomer
- (3) Explain X-ray diffraction method for characterization of polymer.
- (4) Give preparation of Bromamine acid.
- (5) Explain in brief: Determination of  $\alpha$ -Naphthol in brief.
- (6) Give synthesis of Chrome Blue Black R.

(B) Answer any **Three** : **9**

- (1) Give the synthesis of Congo red.
- (2) Explain in brief: Direct determination of amine.
- (3) Write a note on glass transition temperature of polymer.
- (4) Explain IR spectroscopy for analysis of monomer and polymer sample.
- (5) Write a detailed note on melamine formaldehyde with mechanism.
- (6) Write a brief note on paper chromatography.

- (C) Answer any **Two** : **10**
- (1) Discuss Lung nitro meter with diagram.
  - (2) Describe manufacturing of H-acid with diagram.
  - (3) Explain cationic mechanism for addition polymerization of propylene.
  - (4) Explain anionic mechanism for addition polymerization of ethylene.
  - (5) Discuss classification of polymer in detail.
- 3** (A) Answer any **Three** : **6**
- (1) Write types of polyisoprene in detail.
  - (2) Explain homopolymer and copolymer in detail.
  - (3) Write only reaction scheme of neoprene.
  - (4) Explain any one method for the preparation of Anthraquinone.
  - (5) Give the synthesis of Tartrazine.
  - (6) Give preparation of G-acid.
- (B) Answer any **Three** : **9**
- (1) Give synthesis of Naphthol Blue Black 6B.
  - (2) Give synthesis of p-nitro aniline.
  - (3) Give synthesis of Bismark brown.
  - (4) Write a detailed note on nylon,6-6 with mechanism.
  - (5) Discuss manufacturing of polyurethane in detail with reaction scheme.
  - (6) Explain manufacturing of ABS in detail with reaction scheme.
- (C) Answer any **Two** : **10**
- (1) Explain various methods of diazotization.
  - (2) Explain Gas chromatography with schematic diagram.
  - (3) Describe manufacturing of Direct Black EW with diagram.
  - (4) Explain Styrene butadiene rubber in detail.
  - (5) Explain free radical mechanism for addition polymerization of styrene.